

Biochemistry

Edition 8.5

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METABOLISM IN FED AND FASTING STATE

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Well Fed State

00:00:45

AKA Post Prandial State/Absorptive Phase :

- within **2-4 hrs** after food.
- Storage metabolism.
- Components of food digested & absorbed in smaller components.
- ↑ in plasma level of glucose, amino acids, fatty acid, TAGs.
- Hormone of fed state is **insulin**.

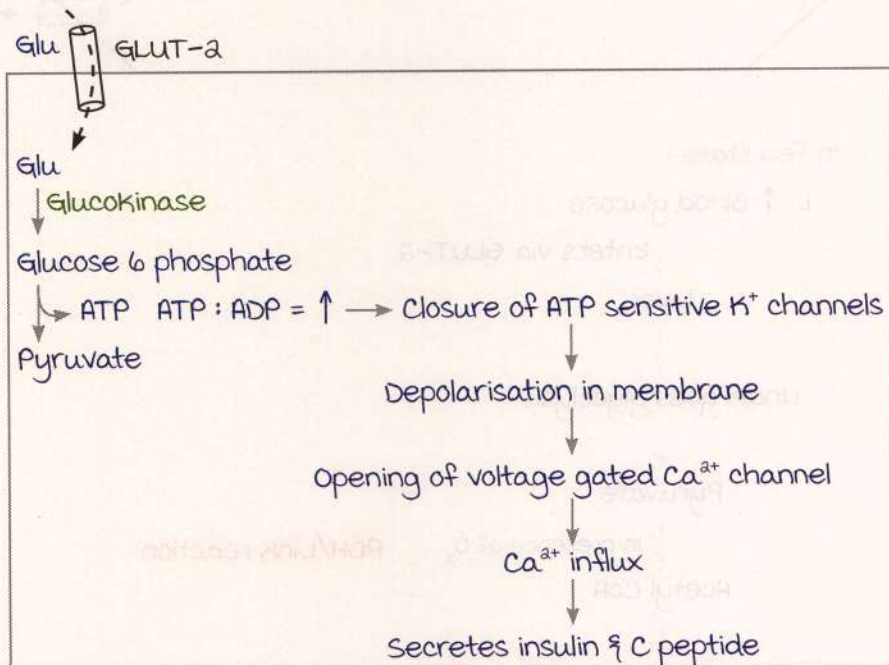
Insulin secretion :

- Begins to rise : At blood glucose $> 70 \text{ mg/dL} / > 3.9 \text{ mmol/L}$.
- Assessment of level of **C-Peptide** = level of insulin.

In Pancreas :

β cells of pancreas :

GLUT-2 : High K_m ,
low affinity for
glucose.



Note :

Insulin is synthesized in **ribosome and rough endoplasmic reticulum**, packaged in golgi apparatus.

Actions of Insulin :

- Favours protein translocation :
 - ↑ GLUT4: Insulin dependent glucose transporter.
 - ↑ Insulin receptor level.

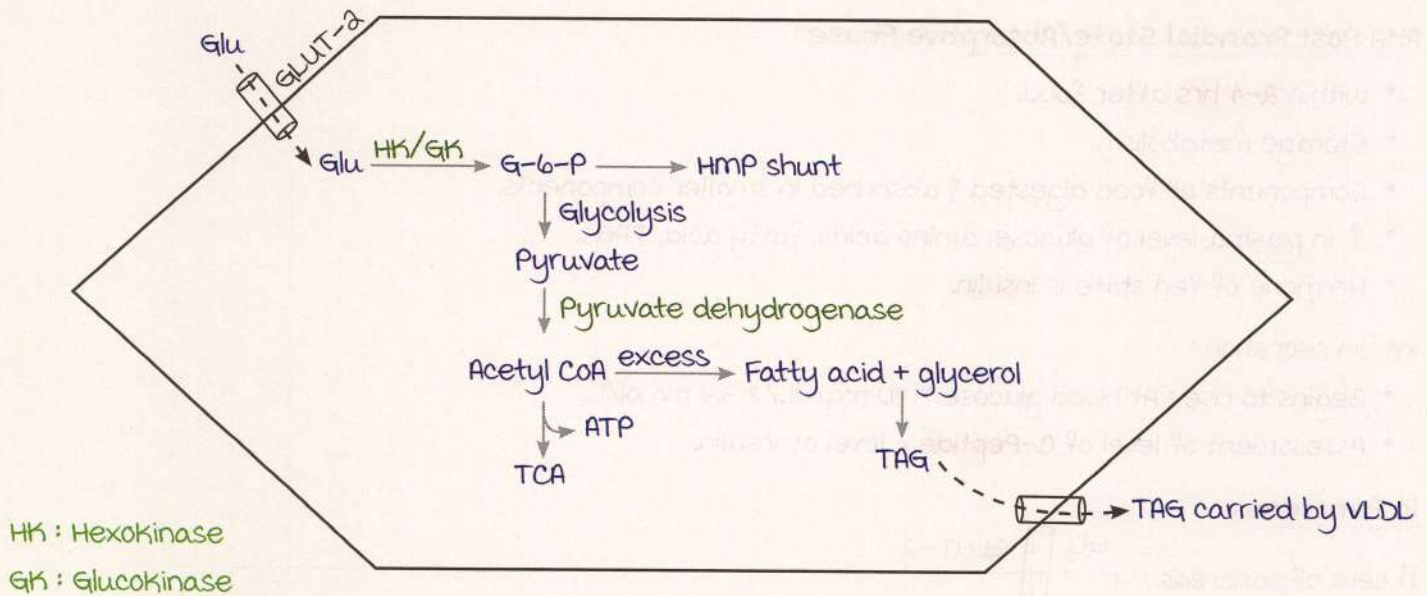
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- ↑ Gene transcription of glucokinase.
- ↑ Enzyme activity : Dephosphorylates regulatory enzyme
 - Phosphodiesterase.
 - Phosphatase.

In Liver :

Liver : Glucose consumer in fed state.

↑ Blood glucose

In fed state :

1. ↑ Blood glucose

↓ Enters via GLUT-2

Liver

undergoes glycolysis

Pyruvate

↓ In presence of O_2

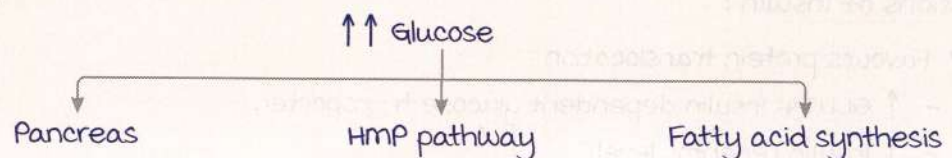
Acetyl CoA

PDH/Link reaction

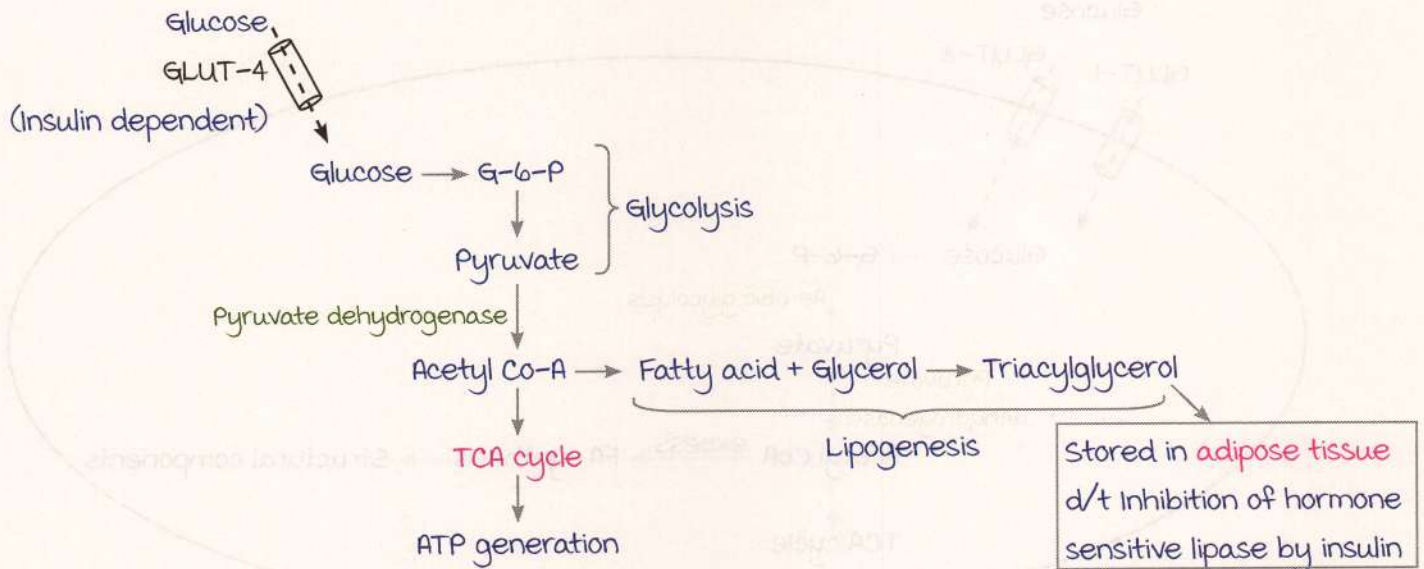
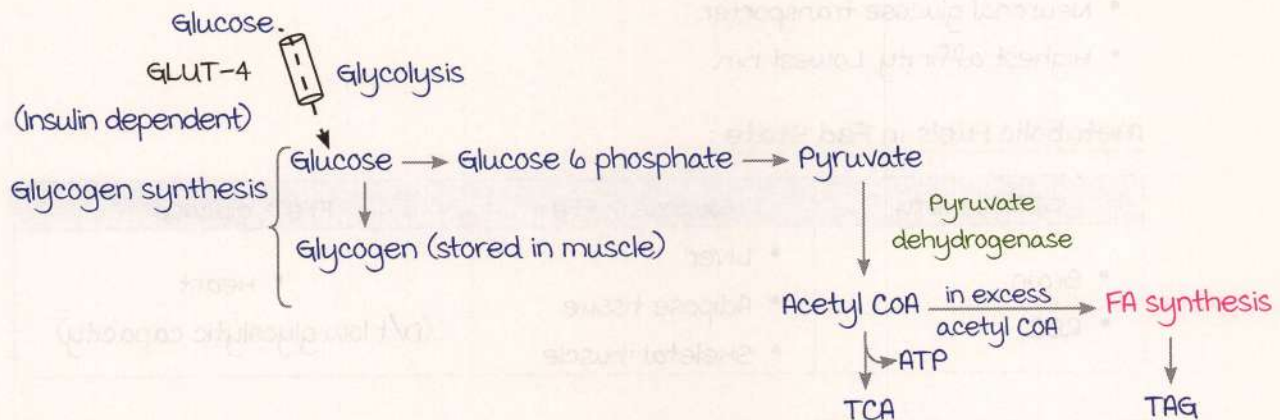
TCA → ATP

2. In excess glucose → Glycogen synthesis : ↑↑ glucose → G6P → Glycogen.

If excess glucose still remains :



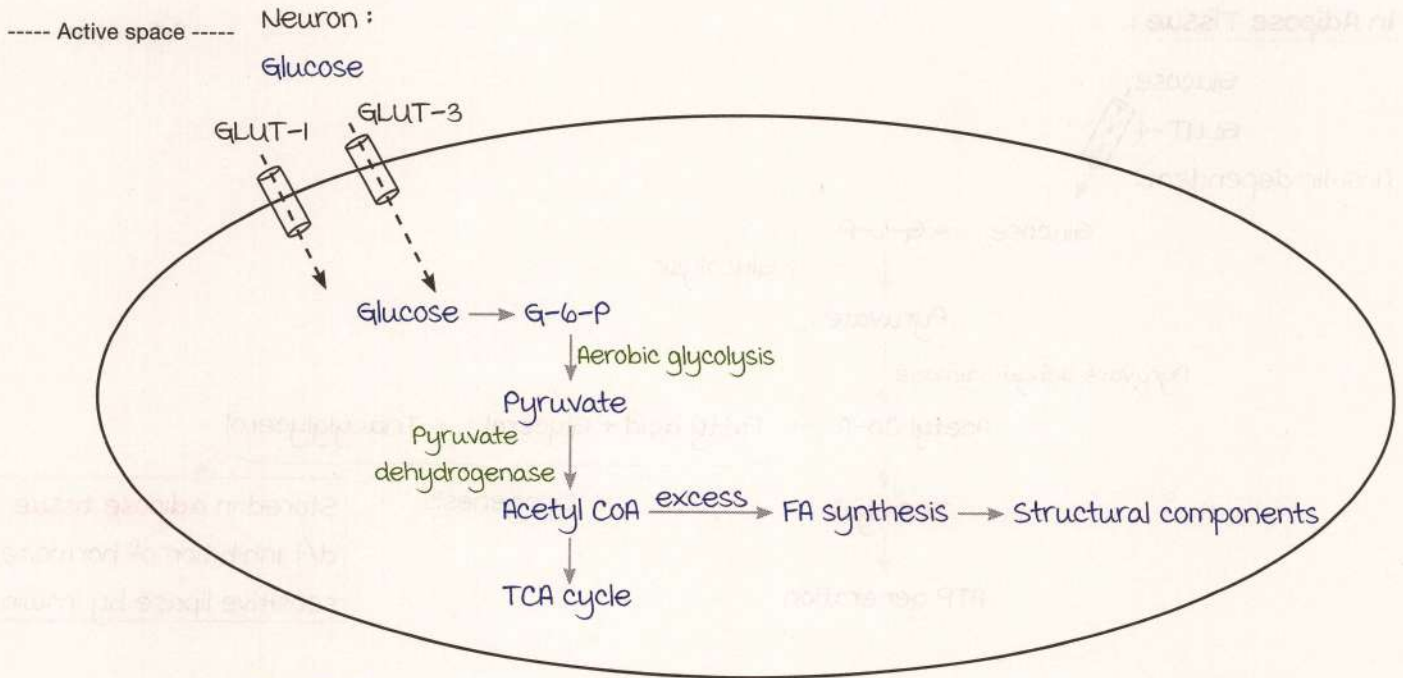
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In Adipose Tissue :In Skeletal muscle :Amino acid in muscles :

- ↑ Protein synthesis
 - Transamination
 - Oxidative deamination
- Removal of amino group → Carbon skeleton → Anabolic functions.

In Brain :Glucose in brain :

- Obligatory requirement of glucose.
- Oxidative pathway only for energy production.



GLUT-3 :

- Neuronal glucose transporter.
- Highest affinity, Lowest K_m .

Metabolic Fuels in Fed State :

Glucose only	Glucose > FFA	FFA > glucose
• Brain • RBC	• Liver • Adipose tissue • Skeletal muscle	• Heart (D/t low glycolytic capacity)

Fasting State/Post Absorptive State

00:28:13

utilizing stored glycogen & triacylglycerol.

Stages	Duration without food intake
Early fasting	4-16 hrs
Fasting	16-48 hrs
Prolonged fasting / starvation	2-5 days
Prolonged starvation	>5 days

Early fasting :

- **Hepatic glycogenolysis** : Source of glucose.
- muscle lacks G-6-Pase → Cannot release free glucose directly.
- Glycogen stores deplete in 16-18 hours.

hepatic
Glycogen $\xrightarrow{\text{glycogenolysis}}$ G-6-P $\rightarrow$ Glucose.

----- Active space -----

Fasting :

Source : Gluconeogenesis $\xrightarrow{\text{ATP (+)}}$

Production of glucose from non-carb substrates :

- Glycerol : From TAG.
- Lactate.
- Alanine : Glucogenic aminoacid from muscle.

Gluconeogenesis :

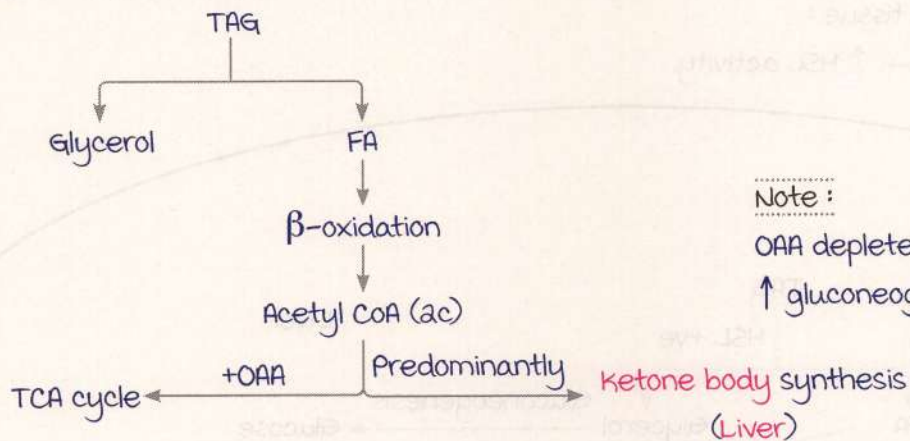
Triacylglycerol (Stored in adipose tissue)

↓ Hormone sensitive lipase $\xleftarrow{(+)}$ Glucagon

Glycerol + FA $\xrightarrow{\beta\text{-oxidation}}$ Acetyl CoA $\rightarrow$ TCA cycle $\rightarrow$ ATP

Starvation/Prolonged fasting :

- ↓ Gluconeogenesis : Non-carb sources are used up.
- Source : Ketone bodies



Note :

OAA depleted d/t

↑ gluconeogenesis in fasting state.

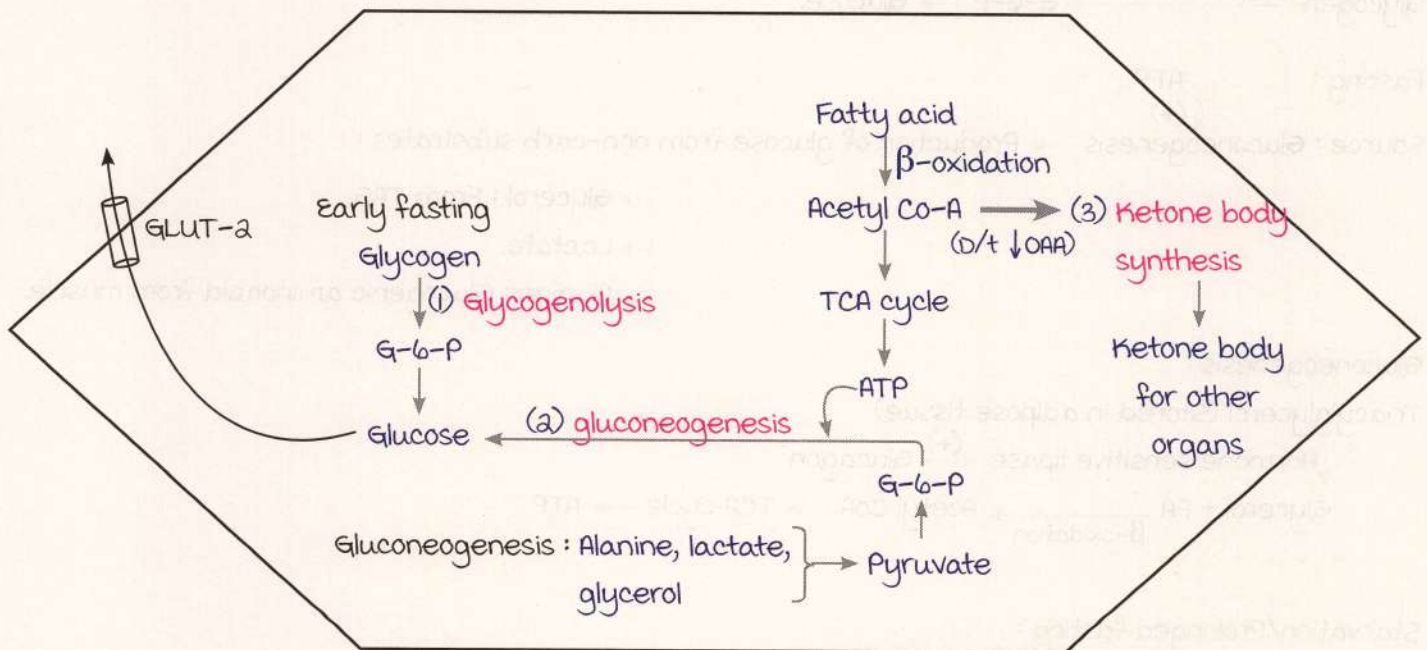
Further Starvation :

Structural muscle proteins
↓ Catabolism
Amino acid
↓
C-skeleton
↓
Catabolic role (Causes cachexia)

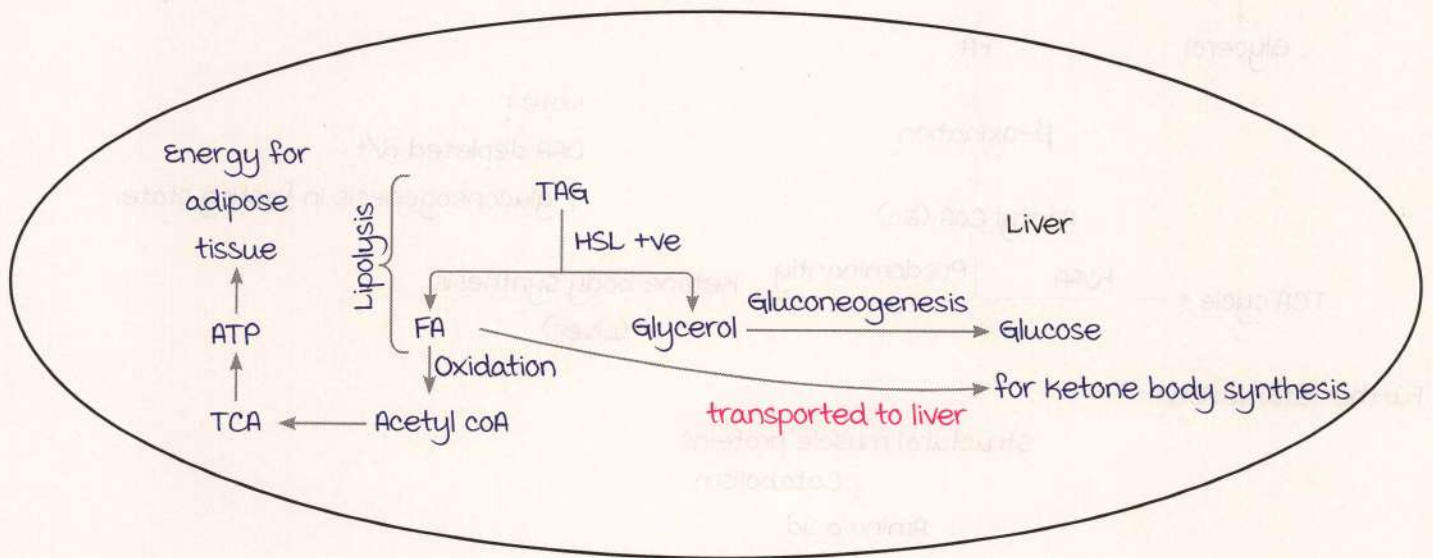
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Fasting Stage in Organs :

In Liver :

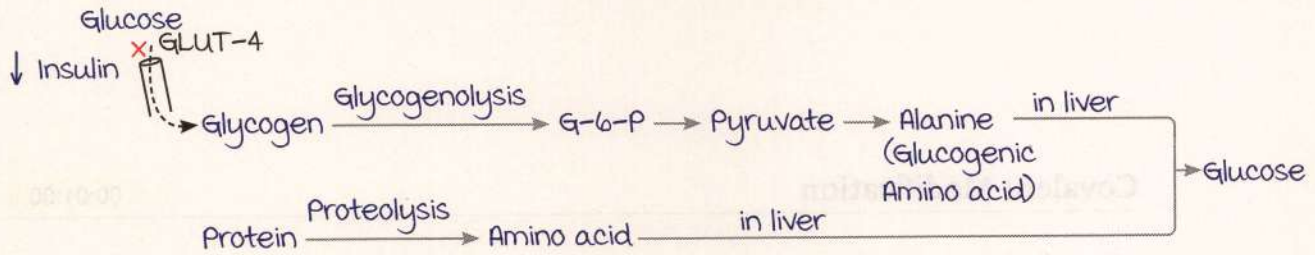


In adipose tissue :

 $\downarrow$ Insulin $\rightarrow \uparrow$ HSL activity.

In skeletal muscle :

----- Active space -----



Source of energy in muscle :

1. **Fatty acid** → Oxidation → Acetyl Co-A → TCA → ATP.
2. **Ketone bodies** from liver.

In brain :

During starvation :

- Available glucose transported via GLUT-3 and converted to ATP.
- Ketone body lysis (provides for only 20% energy requirements) → Acetyl CoA → TCA cycle → ATP.

metabolic fuel in fasting :

Organ	Early fasting/fasting state	Starvation
Brain	Glucose	Glucose/Ketone bodies (20%)
RBC	Glucose	Available glucose Absence of glucose → Lysis of RBC
Liver	Free fatty acid > glucose	Amino acid, FFA, Never ketone bodies
Adipose tissue	FFA > glucose	FFA / Ketone bodies
Skeletal muscle		
Heart		

----- Active space -----

CONCEPT OF ENZYME REGULATION

Covalent Modification

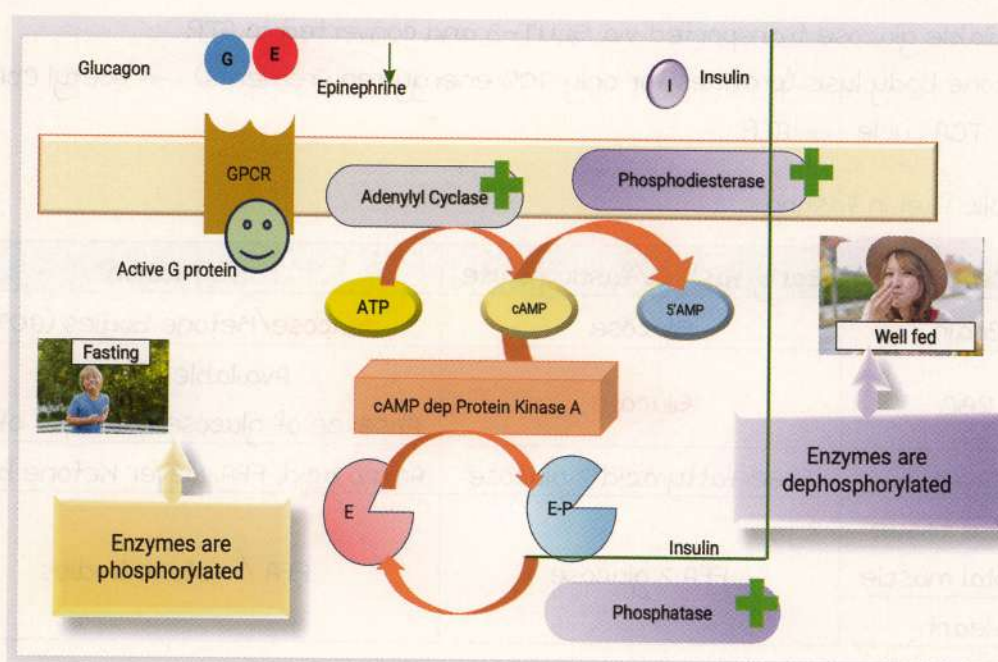
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Sites of phosphorylation

Hydroxyl group containing amino-acids:

- Serine (m/c)
- Threonine
- Tyrosine

mechanism



Examples

Enzyme	Insulin : Glucagon ratio	High activity in
Phosphofructokinase (in glycolysis)	High	DP state
Fructose 1,6 bisphosphatase (in gluconeogenesis)	Low	P state
Glycogen synthase	High	DP state
Glycogen phosphorylase (in glycogenolysis)	Low	P state
Pyruvate dehydrogenase (link between glycolysis & TCA cycle)	High	DP state

Enzyme	Insulin : Glucagon ratio	High activity in
Acetyl CoA carboxylase (Fatty acid synthesis)	High	DP state
HMG CoA reductase (Cholesterol synthesis)	High	DP state
Hormone sensitive lipase (hydrolysis of stored TAG in adipose tissues)	Low	P state

----- Active space -----

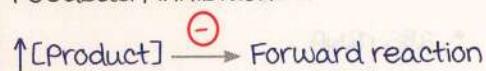
Allosteric Regulation

00:16:31

Feed Forward reaction



Feedback inhibition



Enzyme	Allosteric activator	Allosteric inhibitor
Phosphofructokinase (In glycolysis) Fructose-6-phosphate → Fructose-1,6-bisphosphate (Product)	Substrates in glycolysis • 5' AMP • Fructose-6-phosphate	Products of glycolysis: • ATP • Low pH (d/t lactic acid) • Citrate (formed from acetyl CoA)
Acetyl CoA carboxylase Acetyl CoA → malonyl CoA	Citrate (substrate)	• malonyl CoA (product) • Acyl CoA (fatty acid product)
ALA synthase	-	Heme (product)

----- Active space -----

INTRODUCTION TO ENZYMES

Enzymes :

Specialised proteins that can act as biological catalysts.

Exception :

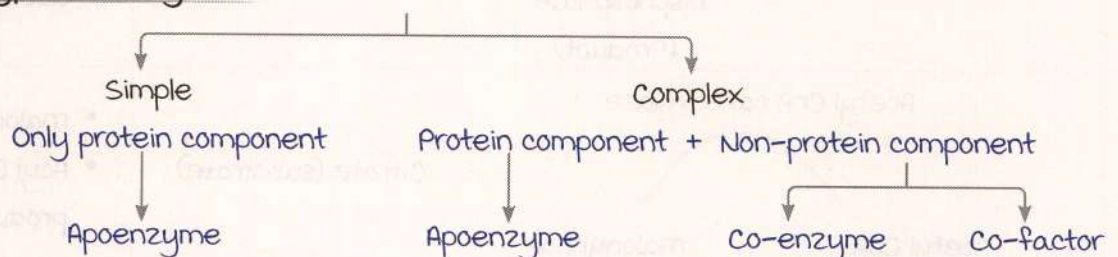
Ribozyme : RNA acts as enzymes.

Ribozymes	Function
Ribosome • 28S rRNA • Peptidyl transferase	Peptide bond synthesis.
Sn RNA	Splicing of exons : post-transcriptional modification of mRNA.
Group II introns	
Ribonuclease P	Post-transcriptional modifications of tRNA.

Enzymes, Co-Enzymes and Co-Factors

00:05:27

Types of Enzymes :



Properties of Enzymes :

Enzymes are proteins.

- Nitrogen : 16% by weight.
- Heat labile.
- Precipitated by protein precipitating agents.

Co-Enzyme :

- Second substrate or co-substrate.
- mostly B-complex vitamins.

Properties of co-enzyme :

----- Active space -----

- Heat stable.
- Low molecular weight organic molecule.

Examples :

	Active form	Reactions involved.
Thiamine (B_1)	Thiamine pyrophosphate (TPP)	• Oxidative decarboxylation. • Transketolase.
Riboflavin (B_2)	Flavin adenine dinucleotide (FAD) FMN	• Oxidative decarboxylation. • Redox reaction : - Complex I of electron transport chain (ETC). - Predominantly FMN.
Niacin (B_3)	Nicotinamide adenine dinucleotide (NAD^+), Nicotinamide adenine dinucleotide phosphate ($NADP^+$)	• Oxidative decarboxylation : Predominantly NAD^+ • Oxidative-reduction reaction (dehydrogenase)
Pantothenic acid (B_5)	Co-enzyme A	Transfer of acyl group
Pyridoxine (B_6)	Pyridoxal phosphate (PLP)	• Transamination. • Trans-sulfuration.
Folic acid (B_9)	Tetrahydrofolate (THFA)	One carbon transfer.
Cobalamin (B_{12})	methyl B_{12}	Homocysteine methyl transferase
	Adenosyl B_{12}	methyl-malonyl CoA mutase
Lipoate	Lipoamide - Oxidised form - Reduced form	oxidative decarboxylation
Ascorbic acid (C)	Ascorbate	Hydroxylation reaction

Co-Factor :

- Inorganic molecules.
- Predominantly minerals.

----- Active space ----- Types

metalloenzyme :

- metal & apoenzyme **tightly integrated**.
- Eg :
 - Cu in tyrosinase.
 - Zn^{2+} in {
 - Carbonic anhydrase.
 - Carboxy peptidase.

metal activated enzyme :

- metal **not tightly integrated** with enzyme.
- Presence of metal is required for enzyme action.
 - Eg: Ca^{2+} required for action of lipase.

Prosthetic Group :

Co-enzyme/ metalloenzyme (Co-factor) tightly integrated to enzyme.

Holoenzyme

00:17:29

- Completed form of complex enzyme.
- **Apoenzyme + co-enzyme/ co-factor.**

Examples :

metals	Enzyme	Function
Zinc	Carbonic anhydrase.	Transport of CO_2 .
	Carboxypeptidase A & B .	Digestion of proteins.
	Alcohol dehydrogenase.	Retinol $\rightleftharpoons$ Retinal (Vision).
	Alkaline phosphatase.	Removal of phosphate in alkaline medium .
	ALA dehydratase.	Synthesis of heme.
	Adenosine deaminase.	Purine catabolism.
	Cytosolic superoxide dismutase (SOD).	Free radical scavenging : Anti-oxidant.
	Lactate dehydrogenase.	Anaerobic glycolysis.
magnesium	• Kinase. • Phosphatase. • mutase. • Enolase.	Transfer of phosphate.
Iron	Heme iron : Complex III & IV of ETC (present in cytochrome).	-
	• Nitric oxide synthase. • Peroxidase, catalase.	• Synthesis of nitric oxide. • Free radical scavengers.

----- Active space -----

metals	Enzyme	Function
Iron	Tryptophan dioxygenase	-
	Non-heme iron : Complex I & II of ETC (found as Fe-S complex)	-
manganese	- Kinase, phosphatase - Arginase - Ribonucleotide reductase - mitochondrial SOD	-
molybdenum	xanthine oxidase	Purine catabolism - End product: uric acid - Deficiency of molybdenum: Hypouricemia.
Potassium	- Pyruvate Kinase - Na⁺-K⁺ ATPase	-
Copper	Tyrosinase	melanin synthesis Deficiency of copper : Hypopigmentation.
	Complex IV ETC	Energy production
	Lysyl oxidase	Collagen synthesis Deficiency of copper: Bleeding manifestations.
Nickel	urease	Not seen in humans
Calcium	- Lecithinase - Lipase	-
Selenium	Glutathione peroxidase	Free radical scavenger (Anti-oxidant)
	Thioredoxin reductase	-
	Deiodinase	Thyroid hormone synthesis
	Selenoprotein P	-

----- Active space -----

CLASSIFICATION OF ENZYMES

Overview :

Trivial name of enzyme :

- Named after reaction mechanism (m/c) / substrate.
- Can be common for 2 enzymes.

IUBMB Classification of Enzyme :

(International Union of Biochemistry and molecular Biology)

Enzyme commission/class/code number : 4 digits.

Eg : 1111 (Alcohol dehydrogenase).

1 st digit :	2 nd digit :	3 rd digit :	4 th digit :
Class	Subclass	Subsubclass	Unique number for every enzyme

Classes of enzymes :

7 Classes → mnemonic : **O**peration **T**heatre **H**as **L**ow **I**ntensity **L**ight

- Oxidoreductase.
- Transferase.
- Hydrolase.
- Lyase.
- Isomerase.
- Ligase.
- Translocase (added on August 2018).

Class I: Oxidoreductases

00:08:07

Enzyme that catalyze oxidation-reduction reactions.

Subclass I : Dehydrogenases (DH)

- Catalyze transfer of hydrogen elements (H^+ , H^- , H_2) & electrons to an acceptor in a coupled oxidation-reduction reaction.

Acceptors (Co-enzymes) :

Flavoproteins :

- $FAD \rightarrow FADH_2$
- Catalyzing enzymes
 - Acyl CoA DH (in β -oxidation).
 - Succinate DH (in Tricarboxylic acid cycle).
 - Glycerol-3-P DH (mitochondrial).